

$$\begin{aligned} 64 \quad \log(5x^2+4) &= 2 \log(3x^2) - \log(2x^2-1) \\ &= \log 9x^4 - \log(2x^2-1) \end{aligned}$$

$$\log(5x^2+4) = \log\left(\frac{9x^4}{2x^2-1}\right)$$

$$(2x^2-1)5x^2+4 = \frac{9x^4}{2x^2-1}$$

$$10x^4 + 8x^2 - 5x^2 - 4 = 9x^4$$

$$-9x^4$$

$$x^4 + 3x^2 - 4 = 0$$

$$(x^2+4)(x^2-1) = 0$$

$$x^2+4=0$$

$$\sqrt{x^2} = \pm 2i$$

$$(x-1)(x+1)$$

$$x = 1, -1$$

$$\sqrt{x^2} = \pm 1$$

$$x = \pm 1$$

$$78 \quad \log_5\left(\frac{x^2}{8}\right) - 3 = \log_5 \frac{x}{40}$$

$$\log_5\left(\frac{x^2}{8}\right) - \log_5\left(\frac{x}{40}\right) = 3$$

$$\log_5\left(\frac{x^2}{8} \cdot \frac{40}{x}\right) = 3$$

$$\log_5(5x) = 3$$

$$5x = 125$$

$$x = 25$$

Natural Log Operations

$$\ln x + \ln(x+3) = 2$$

$$\ln(x^2 + 3x) = 2$$

$$x^2 + 3x = e^2$$

$$x^2 + 3x - e^2 = 0$$

$$x = \frac{-3 \pm \sqrt{9 - 4(1)(-e^2)}}{2(1)}$$

$$x = \frac{-3 \pm \sqrt{9 + 4e^2}}{2} = \frac{1.605}{2}$$

$$\frac{2e^{2x-5}}{2} = \frac{32}{2}$$

$$\ln e^{2x-5} = \ln 16$$

$$2x - 5 = \ln(16)$$

$$\frac{2x}{2} = \frac{\ln(16) + 5}{2}$$

$$x \approx 3.886$$

$$\log 4^{2x+3} = \log 75$$

$$\frac{(2x+3) \cdot \log 4}{\log 4} = \frac{\log 75}{\log 4}$$

$$2x+3 = \frac{\log 75}{\log 4}$$

$$\cancel{2x} = \frac{\log 75}{\log 4} - 3$$

$$\hat{=} \frac{\log 75 - 3 \log 4}{2} \approx 0.057$$

$$e^{2x} + 3e^x = 28$$

$$e^{2x} + 3e^x - 28 = 0$$

$$(e^x + 7)(e^x - 4) = 0$$

$$e^x + 7 = 0 \quad e^x - 4 = 0$$

$$\ln e^x = \ln -7 \quad \ln e^x = \ln 4$$

$$\cancel{x = \ln(-7)}$$

$$x = \ln 4 \approx 1.386$$

Radioactive Iodine has a half-life of 60 days
 It is considered to be safe when 5% or less
 is left. How many days will it take to reach
 a safe level.

$$N = N_0 e^{kt}$$

$$0.5 = 1 e^{k \cdot 60}$$

$$\ln 0.5 = \ln e^{60k}$$

$$\frac{\ln(0.5)}{60} = \frac{60k}{60}$$

$$-0.0116 = k$$

$$N = N_0 e^{kt}$$

$$\ln 0.05 = \ln e^{-0.0116t}$$

$$\frac{\ln(0.05)}{-0.0116} = \frac{-0.0116t}{-0.0116}$$

$$258 \text{ days} = t$$

Newton's Law of Cooling

$$u = T + (u_0 - T)e^{Kt}$$

$$\frac{72}{-71} = \frac{71}{-71} + (75 - 71)e^{K \cdot 1}$$

$$\frac{1}{4} = \frac{4e^K}{4}$$

$$\ln 0.25 = \ln e^K$$

$$-1.386 = K$$

Room Temp = 71°

Normal body temp = 98.6

Body found = 75°

1 hr. later = 72°

$$\frac{75}{-71} = \frac{71}{-71} + (98.6 - 71)e^{-1.386t}$$

$$\frac{4}{27.6} = 27.6e^{-1.386t}$$

$$\ln\left(\frac{4}{27.6}\right) = \ln e^{-1.386t}$$

$$\frac{\ln\left(\frac{4}{27.6}\right)}{-1.386} = \frac{-1.386t}{-1.386}$$

$$1.39 = t$$

$$1.39 \text{ hrs.} = t$$